Measuring SFERA school 2013

LABORATOIRE PROCÉDÉS, MATÉRIAUX et ENERGIE SOLAIRE

.UPR 8521 du CNRS. conventionnée avec l'université de Perpignan

PROCESSES, MATERIALS and SOLAR ENERGY LABORATORY

Emmanuel Guillot CNRS-PROMES Odeillo, France

SEVENTH FRAMEWORK PROGRAMME

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Plan

- Measuring?
- Part I: Instrumentation
- Part II: Uncertainties
- [Part III: Quality]
- Measurement techniques

Special slide

Some tools should be shortly defined and usable here

Introduction

What is measuring?

- Determine a numeric value of a physical parameter in a given set of conditions
 - => instrumentation
 - With an evaluated trust of the numeric value
 - => uncertainties

With an evaluated trust of the procedure => quality

It is a science!

Instrumentation + Uncertainties = **Metrology**

Metrology is defined by the International Bureau of Weights and Measures (BIPM) as "the science of measurement, embracing both experimental and theoretical determinations at any level of uncertainty in any field of science and technology."

Part

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MEASURING IS COMPARING

Measuring: determine a numeric evaluation of a physical parameter of a process

Primary characteristics: time, length, mass...
 Derived characteristics: speed, surface, mass flow, viscosity, specific heat, hardness...

Measuring: determine a numeric evaluation of a physical quantity of a process...

...With comparison to a reference quantity => Number + Unit

What is the length of the car? 4,3 m What is the temperature of the oil? 235 °C How strong is the DNI of the sun? 954 W/m2

MEASURING IS COMPARING

The SI system of units

- 7 units to define it all:
 - Temperature => kelvin
 - Time => second
 - Length => meter
 - Mass => kilogram
 - Luminous intensity => candela
 - Quantity of matter => mole
 - Electric current => ampere

The SI system of units



The SI system of units

Definitions of the units? Universal!

- It should be stable in time

⇒ …

- With a repeatable procedure

- \Rightarrow Second = number of pulsations of transition state of Cesium
- ⇒ Meter = distance travelled by light in vacuum in 1 second
- \Rightarrow Mole = as many as many atoms in 12 mg of Carbon 12

 \Rightarrow Kilogram = mass of the International Prototype Kilogram

The SI system of units

SI = <u>Système International d'unités</u>
French Revolution: Universal for Mankind
=> including the measurement system
=> still many things in French by French organisations





Traceability

MEASURING IS COMPARING

Traceability

International References

National References

Regional / Private References

User Measurements

Comparison: Process of Measurement



Process of measurement

Direct quantity

Width of a rectangle

width





Process of measurement

height

Example of indirect quantity Surface of a rectangle $S = h \times w$ i

Process of measurement



One observation of a measurement

At the end, a numeric evaluation with a unit

The width of the rectangle is 13,45 cm

The surface of the rectangle is 127 cm²

Uncertainties

Partl

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Reference

Guide to the expression of Uncertainty in Measurement



MEASURING IS COMPARING

But how good is the comparison?

How trustworthy is it?

Process of measurement



Uncertainties

Provide a **reasonable** *evaluation* of how much **doubt** we have about the numeric *evaluation* of the measurement

The Truth Is Out There



Uncertainty

Measurement = number + unit + uncertainty

the length of the truck is

12,5 m ± 0,1 m with 95 % confidence

Significance of differences



From WMO — Instruments And Observing Methods Report No. 86

Conformity tests



Modelisation of a measurement

{One observed value}

(True value) + (systematic error) + (random error)

Modelisation of a measurement



Systematic error

If a **systematic error** arises from a **recognized** effect of an influence quantity on a measurement result,

the effect can be quantified and, if it is significant in size relative to the required accuracy of the measurement,

a **correction** or **a correction factor** can be applied to compensate for the effect.

It is assumed that, after correction, the expectation or expected value of the error arising from a systematic effect is zero.

Systematic error

- Examples:
 - While measuring a resistance, the connection wires => $R_{observerd} = R_{unknown} + R_{wires}$
 - The thermal expansion of a ruler => $L = L_0 + \alpha \cdot \Delta T$

A systematic bias observed during calibration of the sensor

Random error

Random error presumably arises from **unpredictable or stochastic temporal and spatial variations** of influence quantities.

The effects of such variations give rise to variations in repeated observations of the measurand.

Although it is not possible to compensate for the random error of a measurement result, **it can usually be reduced by increasing the number of observations**; its expectation or expected value is zero.

Uncertainty evaluation

Systematic errors can be reduced with a correction

=> but we only have an estimate of the correction

Random error can be **reduced** with a **large number** of observations

=> effect of the size of the set on the estimate knowledge??
Uncertainty evaluation

- Method:
 - 1. Describe the measurement: list all the influence quantities
 - 2. Determine each quantity
 - 3. Determine the uncertainty for each quantity
 - 4. Calculate the combined uncertainty
 - . Calculate the expanded uncertainty

Uncertainty evaluation

1. Describe the measurement

Y is determined from N quantities Xi

$$Y = f\left(X_1, X_2, \dots, X_N\right)$$



5Ms — Ishikawa — Fishbone



Uncertainty of the Measurement

Process Environment Material

5M — Ishikawa — Fishbone



Uncertainty evaluation

- Method:
 - 1. Describe the measurement: list all the influence quantities
 - 2. Determine each quantity
 - 3. Determine the uncertainty for each quantity
 - 4. Calculate the combined uncertainty

Calculate the expanded uncertainty

Uncertainty evaluation

3. Determine the uncertainty for each quantity

=> 2 cases:

- Repeated observations => TYPE A
- Other evaluation => TYPE B



Uncertainty Type A

If we have *n* repeated observations:

 \Rightarrow The best *estimate* of the quantity is the **mean**

$$\overline{q} = \frac{1}{n} \sum_{k=1}^{n} q_k$$

 $\Rightarrow \text{ The best estimate of the uncertainty is}$ $u = s_p / \sqrt{n} \quad \text{with} \quad s^2(q_k) = \frac{1}{n-1} \sum_{j=1}^n (q_j - \overline{q})^2$

Uncertainty Type B

If the quantity is not determined from repeated observations, the uncertainty is evaluated by scientific judgement based on all of the available information on the possible variability.

Examples: • manufacturer's specifications

- data provided in calibration and other certificates
- uncertainties assigned to reference data taken from handbooks

Uncertainty Type B

 \Rightarrow Use the existing knowledge

 \Rightarrow Assume a distribution law of the variations

 \Rightarrow Calculate the uncertainty



Uncertainty Type B

For a numeric display ±a

For a hysteresis ±a

$$u(\mu_t) = a / \sqrt{3}$$



Uncertainty evaluation

- Method:
 - 1. Describe the measurement: list all the influence quantities
 - 2. Determine each quantity
 - 3. Determine the uncertainty for each quantity
 - 4. Calculate the combined uncertainty

Calculate the expanded uncertainty

Combined uncertainty

We have the law $Y = f(X_1, X_2, ..., X_N)$

We have the X_i and their uncertainties

 $u = s_{\rm p} / \sqrt{n}$ $u(\mu_t) = a / \sqrt{3}$

=> The combined uncertainty is (uncorrelated quantities)

$$u_{c}^{2}(y) = \sum_{i=1}^{N} \left(\frac{\partial f}{\partial x_{i}}\right)^{2} u^{2}(x_{i})$$

The partial derivatives $\partial f/\partial x_i$ are equal to $\partial f/\partial X_i$ evaluated at $X_i = x_i$

Combined uncertainties

Example:

additive measurement of 2 quantities with equiprobable distributions



Combined uncertainties



Absolute uncertainty NOT relative values

Uncertainty evaluation

- Method:
 - 1. Describe the measurement: list all the influence quantities
 - 2. Determine each quantity
 - 3. Determine the uncertainty for each quantity
 - 4. Calculate the combined uncertainty
 - Calculate the expanded uncertainty

Expanded uncertainty

u(Xi) describes the uncertainty But we would like to say: *the length is 12,5 m* ± 0,1 m with 95 % confidence

- => Expanded uncertainty U
- => Coverage factor k

 $U = k u_{\mathbf{C}}(y)$

Expanded uncertainty

Assuming a few things (normal distributions...) => For 95% confidence **k** = 2

=> For 99% confidence **k** = 3





Assumptions

Normal distributions

Large number of observations

No correlations between quantities

Measurement techniques

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Instrument properties

- Measurement range
- Linearity accuracy of response within range
- Stability short and long term drift
- Response time
- Accuracy
- Precision
- Hysteresis
- Quantization signal and sampling rate
- Cost money, time, complexity

Measurement range

How wide is the possible measurement range?

- Examples:
 - Size of a ruler
 - Starting and destruction speed of an anemometer
 - Freezing and boiling points of a thermometer

Linearity

How many corrections to apply along the measuring range?



Stability

How much drift of the measurement evaluation:

- short term
- long term

Example for a temperature measurement by thermocouple:

- Short term drift: thermal sensitivity of the ADC
 - Long term drift: chemical alteration of the TC

Repeatibility and Reproducibility

Repeatability

Variability on an occasion With-in run precision

Reproducibility

Variability on different occasions Between-run precision

Response time

How fast the output signal changes?



Thermocouples: Speed vs Diameter

Accuracy and Precision

Accuracy (Justesse)

The closeness of the experimental mean value to the true value.

High accuracy = Small systematic error.

Precision (Fidélité)

The degree of scatter in the results. *High precision = Small random error.*

Accuracy and Precision



ACCURATE



whatever...





ACCURATE and PRECISE

Hysteresis

Does the output depends on past environment?







Quantization

Quantity of steps between the analog signal and the numeric value:

- Signal output
- Sampling rate

Eg: 16 bits = 65536 values for the Full Scale of the converter



Eg: 1 ksps = 1000 values per seconds

Instrument properties

- Measurement range
- Linearity
- Hysteresis
- Stability
- Response time
- Quantization
- EXPOSED TYPE



- Accuracy
- Precision
- Repeatability
- Reproductibility
- Cost €€€-time





Choice of the instrument



Instrument properties

There are no perfect sensor which has the perfect properties for all the measurements needs.

⇒Need to adapt the technology and setup of the sensor to the actual requirement of measurement performance: "the size of the uncertainty"

⇒In order to save time and €€€
⇒In order to be realistic with the environment

Instrument properties

There are no perfect sensor which has the perfect properties for all the measurements needs.

⇒A wished performance may be unreachable with the provided resources and the current state of the art of the Metrology

► Eg: measuring the irradiated surface temperature of a tower solar receiver at ±1 K @ 95% uncertainty: next to impossible in real field, at least for now... no ?

Summary LABORATOIRE PROCÉDÉS, MATÉRIAUX et ENERGIE SOLAIRE .UPR 8521 du CNRS. conventionnée avec l'université de Perpignan PROCESSES, MATERIALS **Emmanuel Guillot** and SOLAR ENERGY LABORATORY **CNRS-PROMES** Odeillo, France CNIS SEVENTH FRAMEWORK PROGRAMME PROMES 🔆 UPVD

Measuring is Comparing

The Truth is Out there
Reference books about measurement techniques

 Béla G. Liptak *CRC Press 4th edition: 2005* ISBN13: 978-084-931-0812

Georges Asch
Éditions DUNOD
7th edition: 2010

ISBN13: 978-210-054-9955



7^e édition

DUNOD

THE reference guide for uncertainties, terms





http://www.bipm.org





http://www.bipm.org/en/publications/guides/gum.html

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